

Safety Refresher for Chunsheng Wang Lab

Jessica Snyder Laboratory & Chemical Safety Specialist

True or False?

- My laboratory is a perfectly safe place.
- I know all the hazards of everything I work with.
- I know the hazards of everything other people work with around me.
- I know who to talk to if I have a safety concern.
- I am expected to talk to others about safety concerns.
- I am responsible for safety.



Why is this important?

- Your laboratory is a high hazard work area
 - Highly reactive chemicals
 - Compressed gases (flammable, oxidizers, and inert)
 - Electrical equipment
- Safe science is good science!
 - Your health and safety are most important
 - What can harm you can harm your research
 - Injuries cost you time to heal
 - Accidents destroy experiments, equipment, and supplies
 - Accidents injure your reputation, the lab's reputation, and UMD's reputation.

What are *your* responsibilities for conducting safe research?

- Practice good laboratory safety
- Understand hazards associated with your research
- Understand how to respond to incidents
- Ensure laboratory waste is managed properly
- Keep your areas clean and organized





DEPARTMENT OF ENVIRONMENTAL SAFETY, SUSTAINABILITY & RISK

Chemical Hygiene

Recognizing chemical hazards

- Chemical Labels
- Safety Data Sheets
- Reference materials
 - > Prudent Practices in the Laboratory
 - PubChem (CDC, online)
 - Cameo Chemicals (online)
- Dr. Wang
- Senior Lab Members
- Standard Operating Procedures
- Internet search





SDS: Section 10

Section 10 **Stability and Reactivity** contains:

- Incompatible materials
- Decomposition products
- Other hazardous conditions and reactions.

10. STABILITY AND REACTIVITY

10.1 Reactivity

no data available

10.2 Chemical stability

Stable under recommended storage conditions.

10.3 Possibility of hazardous reactions

no data available

10.4 Conditions to avoid

no data available

10.5 Incompatible materials

Bases, Amines, Alkali metals, Metals, permanganates, e.g. potassium permanganate, Fluorine, metal acetylides, hexalithium disilicide

10.6 Hazardous decomposition products

Other decomposition products - no data available In the event of fire: see section 5

This is Section 10 from the SDS for concentrated hydrochloric acid from Sigma-Aldrich.



GHS Hazard Symbols

Hazard symbols or pictograms convey information about the health, physical and environmental hazards associated with the chemical. There are nine pictograms.



Acute Toxicity (Severe)



Acute Toxicity (Harmful)
Dermal Sensitizer,
Narcotic Effects,
Respiratory Tract Irritant



Carcinogen
Respiratory Sensitizer
Reproductive Toxicity
Target Organ Toxicity
Mutagenicity
Aspiration Toxicity



Corrosive Severe Skin Burns and Eye Damage, Serious Eye Damage



Flammables, Self-Reactives, Pyrophorics, Self-Heating, Emits Flammable Gas, Organic Peroxides



Gases Under Pressure



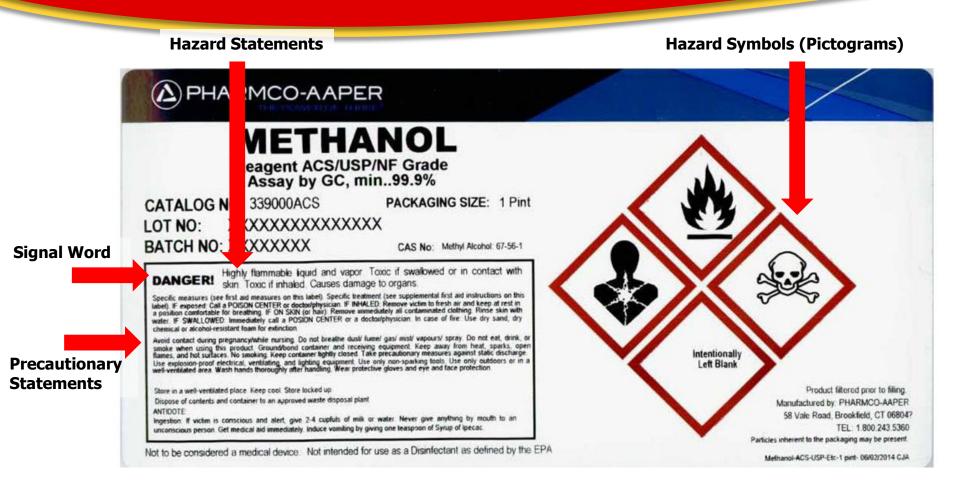
Explosives, Self-Reactives, Organic Peroxides



Oxidizers



GHS Label



GHS Safety Data Sheets

Safety data sheets (SDS) communicate the hazards associated with the product and provide guidelines for safe handling and storage.

Each SDS has 16 sections. We will be looking at 4 of these sections in detail:

- Section 2: Hazards Identification
- Section 3: Composition/Information on Ingredients
- Section 8: Exposure Controls/Personal Protection
- Section 10: Stability and Reactivity



SDS: Section 2

Section 2: Hazards Identification contains

- GHS health hazard classifications
- Hazard statements
- Precautionary statements
- Pictograms

2. HAZARDS IDENTIFICATION

2.1 Classification of the substance or mixture

GHS Classification in accordance with 29 CFR 1910 (OSHA HCS)

Oxidizing liquids (Category 3), H272 Skin corrosion (Category 1), H314 Serious eye damage (Category 1), H318

For the full text of the H-Statements mentioned in this Section, see Section 16.

2.2 GHS Label elements, including precautionary statements

Pictogram

Signal word

Hazard statement(s)

H272 H314 H318

Precautionary statement(s)

P210 P220 P221 P264 P280

> P301 + P330 + P331 P303 + P361 + P353

Aldrich - 433233

May intensify fire; oxidizer.

Causes severe skin burns and eye damage. Causes serious eye damage.

Keep away from heat.

Keep/Store away from clothing/ combustible materials.

Take any precaution to avoid mixing with combustibles.

Wash skin thoroughly after handling.

Wear protective gloves/ protective clothing/ eye protection/ face protection.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

IF ON SKIN (or hair): Take off immediately all contaminated clothing.

Section 2 from the SDS for concentrated nitric acid (Sigma-Aldrich).



Chemical Storage

- Chemicals should be stored according to hazard class
 - Not Alphabetically
 - Some chemicals have multiple hazards (e.g., glacial acetic acid)
- All chemicals should have compatible secondary containment
- Properly label all chemicals
 - Including samples
- Must have appropriate storage location
 - Special Cabinet (e.g., flammable liquids storage cabinet)
 - Not in fume hood
 - Not under sink



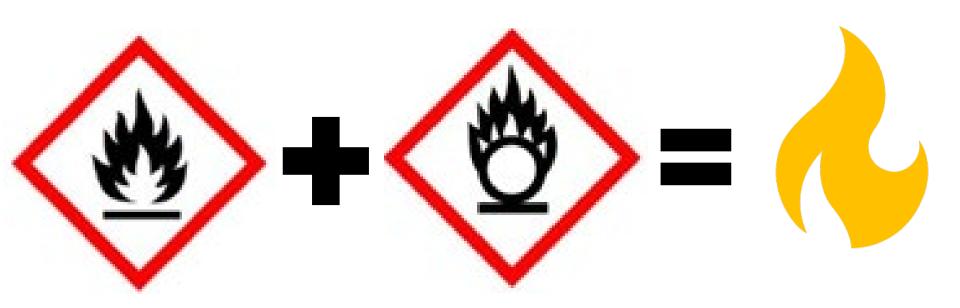


Chemical Storage: Hazards

- Physical Hazards come first
 - Oxidizer
 - > Flammable
 - Corrosive
 - Water or air reactive
 - Explosive
 - Reactivity
- Incompatible classes must be kept separate
 - Most common improper storage: oxidizers and flammables together
- Use location or secondary containment to keep incompatible chemicals separated



Incompatible!



Chemical Storage: Hazards

- Health hazards come second
 - ➤ Toxic
 - Harmful
 - Sensitizer
 - Corrosive
 - Separate toxic chemicals from reactive and corrosive chemicals
- Do not store above eye level



Chemical Storage: Hazards

- Special storage needs:
 - > Flammables:
 - Flammable cabinets or Flammable-rated refrigerators/freezers
 - Away from heat sources and spark hazards
 - Away from oxidizers



- In labeled corrosive cabinets
- Separated from incompatible corrosives
 - Acid/base
 - Oxidizers







Managing Waste

Managing Waste: Why?

- Waste management is mandated by:
 - Federal laws (EPA)
 - Maryland laws (MDE: stricter than EPA)
 - Prince George's county
 - Washington Suburban Sanitary Commissions
- Waste mismanagement has repercussions!
 - Accidental reactions and exposures
 - Fines from regulatory agencies
 - Lab shutdowns



Storage of Chemical Waste

- Waste containers
 - Must be compatible with chemical
 - Must be in secondary containers (bins)
- Segregate waste
 - By hazard
 - From incompatible materials
- Must be labeled fully in English on green tag



Above: properly segregated, contained waste.

Left: A waste bottle melted and sprayed waste when incompatible waste were combined.



Waste Handling

- NEVER mix unknown or incompatible waste
 - If you are not sure, do not mix it
- Waste must be closed at all times unless:
 - It is reacting or off-gassing
 - New waste is being added
- If waste is reacting or generating gas:
 - Keep it open in a secondary container until fully reacted
 - Keep in a fume hood (or glove box) toward the back
 - Place labels and signs indicating what it is, and hazards involved:
 - Caution! Reacting waste: do not close cap or move from hood. Toxic gas!



Above: A waste bottle melted and sprayed waste when incompatible waste were combined.



Managing Waste: Your laboratory



Managing Waste: Your laboratory



Laboratory Safety Contacts

Lab Safety Group: <u>labsafety@umd.edu</u> **Miriam Sharp,** Lab Safety Manager
msharp@umd.edu (301) 405-2070

Jonathan Robertson, Senior Industrial Hygienist (hazardous exposures, chemical & general lab safety) jrober14@umd.edu (301) 405-8755

Jessica Snyder, Industrial Hygienist (chemical & general lab safety)
jasndyer@umd.edu (301) 405-3219

Dmitry Akmal, Research Safety Specialist (electrical and engineering, general lab safety questions) dakmal@umd.edu (310) 405-6087

Hazardous waste questions? Emergency Spills? Environmental Affairs: (waste and spills) envaffairs@umd.edu

